

(19) World Intellectual Property
Organization
International Bureau



(43) International Publication Date
21 October 2004 (21.10.2004)

PCT

(10) International Publication Number
WO 2004/089850 A2

(51) International Patent Classification⁷: C04B 35/573,
41/50, 35/52, C22C 26/00

(21) International Application Number:
PCT/EP2004/050367

(22) International Filing Date: 26 March 2004 (26.03.2004)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
0301117-8 14 April 2003 (14.04.2003) SE

(71) Applicant (for all designated States except US): SKELE-
TON TECHNOLOGIES AG [CH/CH]; Zugerstrasse 72,
CH-6340 Baar (CH).

(72) Inventors; and

(75) Inventors/Applicants (for US only): SVENDSEN,
Lena [SE/SE]; Zinkens väg 65, S-117 41 Stockholm (SE).
ZHENG, Jie [SE/SE]; Hagalundsgatan 14, S-169 63 Solna
(SE). MEURLING, Fredrik [SE/SE]; Wollmaryskulls-
gatan 11, S-118 50 Stockholm (SE). ROSTVALL, Tomas
[—/SE]; Selmedalsringen 8, S-129 36 Hägersten (SE).

(74) Agents: ALBIHNS STOCKHOLM AB et al.; P.O. Box
5581, Linnégatan 2, S-114 85 Stockholm (SE).

(81) Designated States (unless otherwise indicated, for every
kind of national protection available): AE, AG, AL, AM,
AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN,
CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,
GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE,
KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD,
MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG,
PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM,
TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM,
ZW.

(84) Designated States (unless otherwise indicated, for every
kind of regional protection available): ARIPO (BW, GH,
GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW),
Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), Euro-
pean (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR,
GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK,
TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW,
ML, MR, NE, SN, TD, TG).

Published:

— without international search report and to be republished
upon receipt of that report

For two-letter codes and other abbreviations, refer to the "Guid-
ance Notes on Codes and Abbreviations" appearing at the begin-
ning of each regular issue of the PCT Gazette.

(54) Title: A METHOD FOR MANUFACTURING A DIAMOND COMPOSITE

(57) Abstract: The present invention relates to a method for manufacturing a diamond composite, comprising the steps of: a) mixing diamonds with additives, the mixture comprising at least 50 wt% and less than 95 wt% of diamonds and more than 5 wt% additives; b) forming a work piece from the mixture using a pressure of at least 100 Mpa; c) heating the formed work piece to at least 300°C for removing possible water and wholly or partially removing additives; d) heating the work piece and controlling the heating temperature and heating time so that a certain desired amount of graphite is created by graphitization of diamonds, wherein the amount of graphite created by graphitization is 3-50 wt% of the amount of diamond; e) infiltrating silicon or silicon alloy into the work piece, the infiltration of silicon or silicon alloy being performed at a temperature below 1900°C and a pressure less than 50 bars; and f) heating the infiltrated work piece to form silicon carbide, or other carbides, (and silicides), thereby creating a final diamond composite.

WO 2004/089850 A2

J609 Rec'd PCT/PTO 14 OCT 2009

1

A method for manufacturing a diamond composite.

FIELD OF INVENTION

The present invention relates to a method for manufacturing a diamond
5 composite.

BACKGROUND OF THE INVENTION

There is a general need for extremely hard materials in many fields of
application. These applications may be as tools for cutting, turning, milling,
10 drilling, sawing or grinding operations etc. In WO 99/12866 and WO
00/18702 methods using controlled graphitization of diamonds are known for
manufacturing diamond composite materials having excellent properties for
such applications. There is also a need for a material with extreme thermal
conductivity to improve resistance to thermal chock, wear resistance and also
15 for cooling electronic components like microprocessors. In WO 02/42240 the
use of such method is known for manufacturing a composite material having a
high thermal conductivity.

The object of the present invention is to improve such a method in such a way
20 that the forming of the composite is facilitated, thereby enabling bodies having
complex and very precise forms to be manufactured. The object of the
invention is also a material produced by the method with improved
performance.

SUMMARY OF THE INVENTION

The object of the invention is accomplished by a method for manufacturing a
diamond composite, comprising the steps of;

- a) mixing diamonds with additives, the mixture comprising at least 50 wt % and less than 95 wt% of diamonds and more than 5 wt% additives,
- b) forming a work piece from the mixture using a pressure of at least 100 MPa,
- 5 c) heating the formed work piece to at least 300°C for removing possible water and wholly or partially removing additives,
- d) heating the work piece and controlling the heating temperature and heating time so that a certain desired amount of graphite is created by graphitization of diamonds, wherein the amount of graphite created by graphitization is 3-50 wt% of the amount of diamond,
- 10 e) infiltrating silicon or silicon alloy into the work piece, the infiltration of silicon or silicon alloy being performed at a temperature below 1900°C and at a pressure less than 50 bars, and
- f) heating the infiltrated work piece to form silicon carbide, or other carbides (and silicides), thereby creating a final diamond composite..
- 15

In a preferred embodiment the additives in the mixture comprise binders and/or dispersing agents and/or low-friction agents. Advantageously, the mixture is agglomerated in order to facilitate forming of the work piece. The forming of the work piece is preferably made by mechanical pressing, injection moulding or roll compaction or other powder metallurgical forming processes. TiC, BC or SiC-powder can be included in the mixture. The forming of the work piece is made at a pressure of at least 100 MPa, preferably 300-700 MPa, most preferably about 600MPa and the diamond content of the formed work piece is at least 40 vol%, preferably 70-80 vol%.

20

25

The heating of the formed work piece for removing water and additives is preferably made at a mechanical pressure of at least 500 Pa, preferably at least

3000 Pa. This pressure can typically be applied in one direction with a flat weight.

5 More silicon or silicon alloy than the amount needed for forming carbide from the created graphite and filling the pores of the formed carbide skeleton, can preferably be provided so that the surplus of silicon or silicon alloy will form a surface coating on at least one surface of the manufactured diamond composite.

10 The present invention also relates to the use of the inventive method for forming a substrate for a diamond film and to a diamond composite manufactured by the inventive method, wherein at least one side of the composite is coated with a layer of aluminium nitride.

15 DESCRIPTION OF EMBODIMENTS

The method according to the present invention aims to improve the methods presented in WO99/12866 and WO00/18702. These methods include the following steps:

- 20 1. Forming a work piece out of a mixture containing diamonds.
2. Heat treating the work piece and controlling the heating temperature and heating time so that a certain desired amount of graphite is created by graphitization of diamonds.
3. Infiltrating melted silicon or alternatively a silicon alloy into the heated
25 work piece.
4. Reacting of the molten silicon or silicon alloy and graphite to form SiC. When silicon alloy reacts with graphite also other carbides and possibly silicides may form.

By the manufacturing process described above, an article with a predetermined shape is formed. In the present invention, steps similar to the steps 1-4 are used. The main difference from the known methods is the composition of starting materials and the use of high pressure during the forming step.

In a preferred embodiment, the starting material is comprised of diamonds and additives. The diamonds in the mixture preferably consist of at least two different fractions with different diamond particle sizes. To obtain high thermal conductivity at least 50 weight% of the diamond content in the work piece should preferably have a size of 80 μm or above. The use of at least two different fractions of diamonds with different particle sizes is advantageous in order to reach a packing degree in the work piece that in the produced diamond composite gives a high enough diamond concentration (e.g. a short path for the phonons to travel between the diamonds) to reach high levels of thermal diffusivity and thermal conductivity. High diamond content is normally also advantageous for wear resistance. The content of diamond in the work piece is at least 50 wt%, other contents being additives. In order to facilitate forming of the work piece and obtain improved mechanical properties in the final diamond composite in relation to a material manufactured in accordance with the known methods referred to above, the content of additives in the starting material mixture according to the present invention is more than 5 wt%. The additives in the mixture can comprise one or more of binders, dispersing agents, and low-friction agents commonly used in powder metallurgy and also other additives, like TiC, for facilitating the forming of the work piece, facilitating the infiltration of silicon or silicon alloy in the formed work piece and improving the properties of the final composite. The mixture can be made homogeneous in a liquid with the help of dispersing

agents. In order to further facilitate forming of the mixture, the mixture is advantageously agglomerated in a liquid, like water or alcohol.

5 Example of a suitable binder is polyethyleneglycol (PEG). Examples of suitable dispersing agents are ammonium salt of polycarboxylate and steric acid is a suitable low-friction agent. Other known binders, dispersing agents and low-friction agents can also be used.

10 Examples of further additives that can be used are TiC, BC and SiC.

Forming of the work piece is carried out by mechanical pressing, injection moulding, roll compaction or other known powder metallurgical forming methods. The forming should be performed at a forming pressure of at least 100 MPa. Advantageously a pressure of at least 300 MPa - 700 MPa is used.

15 The amount of additives in the mixture should be more than 5wt%. One reason for this is to facilitate the distribution of the agglomerated mixture and to obtain a homogenous body while using high forming pressures. Especially low-friction agents help the redistribution of the components of the mixture under high pressure. Also the binder component should be chosen to limit the

20 internal friction during the redistribution. The function of the binder is otherwise to keep the accurate shape of the work piece during handling and first stages of heating. A work piece formed in such a way will have a very precise form. Furthermore, the high forming pressure increases the packing degree of the diamonds thereby increasing the diamond content per volume

25 and thereby the thermal conductivity and hardness of the produced diamond composite. After the forming step the diamond content in the work piece is at least 40 vol%, preferably 70-80 vol%. At high pressures the diamonds will be more and more crushed and pressures much higher than 700 MPa should normally not be used. The most preferred forming pressure is 600 MPa.

Another advantage of using high forming pressures is that the deformation of the produced diamond composite will be less than with the known methods referred to above, although the deformation in diamond composites produced by these method is low and can be tolerated if not composite specimens with a very precise form are required. It has been shown that the deformations of a composite material specimens produced by the known methods referred to above are at least twice as high as the deformations of a composite material produced according to the present invention. Such deformations include warpage, swelling and shrinking.

After forming the work piece, the work piece is slowly heated to at least 300°C. Water and additives that are volatile or form gaseous compounds at the chosen temperature are thereby removed from the formed work piece. In order to minimize deformation of the work piece, a mechanical pressure of at least 500 Pa, preferably at least 3000 Pa, is optionally applied to the work piece during at least a part of this heating step. Typically this force is applied from one direction with a very flat weight to enhance the flatness of simple or complicated plates.

Thereafter the work piece is heated at a temperature between 1000-1900°C. The total duration of the heat treatment of the work piece is as long as is needed for creating the desired degree of graphitization of the diamonds. The amount of graphite created by graphitization of diamonds should be 3-50 wt% of the amount of diamonds. The heat treatment is made in vacuum or an inert atmosphere and at an atmospheric pressure less than 50 bars. For the properties of the final material it is crucial to obtain graphite created by graphitization of diamonds. Some residual carbon from incompletely removed additives may contribute as a carbon source.

The infiltration of molten Si or silicon alloy is carried out by such known methods as melting a solid piece on or near the surface of the work piece, feeding already molten Si or silicon alloy on to the surface of the work piece or by dipping the work piece into a melt of Si or silicon alloy. As the melt infiltrates the work piece it reacts with graphite and possibly residual carbon and forms SiC or a SiC phase including elements from the alloying elements. The formed phase and a small amount of un-reacted silicon or silicon alloy phase fill up the porous space of the work piece. When additives, such as TiC, BC or SiC, are included these can react with molten silicon or silicon alloy and form a phase or phases including at least one of the elements from the additives. Preferably, a surplus of silicon or silicon alloy is used so that at least one surface of the diamond composite produced is coated with a layer of silicon or silicon alloy.

The heating and the infiltration steps can advantageously be carried out in the same furnace and it is possible to perform the heating steps in one heating cycle. It is of course also possible to perform the heating steps as separate heating operations, for example let the heated work piece cool before reheating the work piece together with silicon or silicon alloy.

The infiltrating melt used can be a silicon alloy comprising at least one metal from the group consisting of Ti, Zr, Hf, V, Nb, Ta, Cr, Mo, W, Mn, Re, Fe, Co, Ni, Cu, Ag, Al, and the elements B or Ge. In this case small amounts of secondary phase compounds may form, such as metal silicides, metal carbides, etc.

The sintered composite material thus consists of three major phases, a diamond phase, a carbide phase around each diamond and un-reacted silicon or silicon alloy phase between areas of carbide. The carbide that has formed

from the reaction between the graphitized diamond and the melt is coating and surrounding each individual diamond particle. The carbide phase forms an interconnected skeleton structure, which is enclosing the diamond particles. There is very little diamond-diamond contact in the composite material. Due to the fact that the carbide has a thermal expansion coefficient larger than the thermal expansion coefficient of diamond, the carbide strives to contract more than the diamonds during cooling after the termination of the heat treatment. At temperatures below the temperature at which the carbide was formed, the carbide surrounding the individual diamond particles will exert a compressive force on each particle. This is believed to contribute to some extent to the surprisingly good thermal conductivity of the material according to the invention. The un-reacted silicon- or silicon alloy together with the possible small amounts of secondary phase compounds, such as metal carbides, metal silicides, etc, are mainly located in the areas in-between the silicon carbide that enclose the diamonds.

To provide a high value of thermal conductivity to a composite material it is crucial to have good adhesion between the different phases, see Handbook of industrial diamonds and diamond films, page 184. The graphite layer formed on the surface of the diamond particles has very good adhesion to the diamond since the graphite is transformed diamond. When the silicon or silicon alloy melt reacts with the said graphite the carbide formed inherits the very good adhesion to the diamond and a strong bond between the carbide and diamond is formed. When nucleation of carbide takes place on a graphite surface that has been formed through graphitization of diamond the formed carbide grows epitaxially, i.e. the growth of carbide on the diamond follows the crystallographic orientation of the diamond. The manner in which carbide is formed and the strong bond between the diamond particles and the surrounding carbide are believed to be decisive factors for obtaining the

surprisingly high thermal conductivity of a material according to the present invention. A long free path for the phonon transport is obtained in the material according to the present invention. The good bond between the diamonds and the carbide phase also has very good strength. If the produced composite is
5 broken, the breakage often goes through the diamonds instead of breaking this bond, this showing the strength of the bond. In order to ensure that each diamond is surrounded by graphite and that a continuous carbide skeleton is obtained, the graphitization of diamond should always be more than 3 wt%, preferably more than 6 wt% of the diamond.

10

The graphitization transforms defective layers on the diamond surface, resulting in improvement of the phonon transport path. Surprisingly it has been found that a direct bonding between diamonds is not needed to achieve good thermal conductivity. To have a phonon transport path of high quality is
15 more essential.

15

It has surprisingly been found that the high pressure used during the forming step, does not hinder the infiltration of silicon or silicon alloy. It has been shown that silicon can be infiltrated into work pieces having porosity well
20 below the lower limit of 25 vol% stated in WO99/12866 and WO00/18702. The present successful method has been achieved using a mixture of diamonds with at least one fraction containing diamonds larger than 0.080 mm. It has been observed that molten silicon is sucked into the work piece. It is believed that this phenomena is due to the size of the larger pores in the powder
25 mixture of diamonds and additives being heavily reduced by the forming made at a high pressure, so that a large percentage of the pores have a capillary size after the forming and the graphitization steps. Such a reduction of the pore volume needed for the infiltration, makes it possible to produce diamond

composites having a very high content of diamond and a low content of un-reacted silicon or silicon alloy.

5 Surprisingly, it has also been observed that powders of TiC in the mixture further facilitates the infiltration of silicon. It has also been shown that TiC improves the mechanical properties of the composite. Good results have also been accomplished with BC and SiC.

10 It has been shown that a diamond composite made by this method is an excellent substrate for a pure diamond layer, made by e.g. Chemical Vapour Deposition, CVD. The high thermal conductivity and the low thermal expansion make a good match with pure diamond layer, both in the CVD process and in electronic applications and wear applications.

15 A diamond composite manufactured by the described method can be coated with a thin layer of aluminium nitride. With one side coated with a 0.001 mm layer of aluminium nitride on diamond composite made in accordance to the present invention, the electrical resistance has be shown to increase 100 times or more. This is important in certain designs of microprocessor packages.

Claims

1. A method for manufacturing a diamond composite, comprising the steps of;
 - 5 a) mixing diamonds with additives, the mixture comprising at least 50 wt% and less than 95 wt% of diamonds and more than 5 wt% additives,
 - b) forming a work piece from the mixture using a pressure of at least 100 MPa,
 - c) heating the formed work piece to at least 300°C for removing possible
10 water and wholly or partially removing additives,
 - d) heating the work piece and controlling the heating temperature and heating time so that a certain desired amount of graphite is created by graphitization of diamonds, wherein the amount of graphite created by graphitization is 3-50 wt% of the amount of diamond,
 - 15 e) infiltrating silicon or silicon alloy into the work piece, the infiltration of silicon or silicon alloy being performed at a temperature below 1900°C and a pressure less than 50 bars, and
 - f) heating the infiltrated work piece to form silicon carbide, or other carbides, (and silicides), thereby creating a final diamond composite.
- 20 2. The method according to claim 1, wherein the additives in the mixture comprise binders and/or dispersing agents and/or low-friction agents.
3. The method according to claim 1 or 2, wherein the mixture is
25 agglomerated in order to facilitate forming of the work piece.
4. The method according to claim 1, 2 or 3, wherein the forming of the work piece is made by mechanical pressing, injection moulding or roll compaction.

5. The method according to any one of claims 1-4, wherein particles of TiC, BC or SiC are included in the mixture.
- 5 6. The method according to any one of claims 1-5, wherein the forming of the work piece is made at a pressure of at least 300 MPa, preferably below 700 MPa.
- 10 7. The method according to any one of claims 1-6, wherein the diamond content of the formed work piece is at least 40 vol%, preferably 70-80 vol%.
- 15 8. The method according to any one of claims 1-7, wherein the removing of water and additives by heating the formed work piece is made at a mechanical pressure of at least 500 Pa, preferably at least 3000 Pa.
- 20 9. The method according to any one of claims 1-8, wherein more silicon or silicon alloy than the amount needed for forming carbide from the created graphite and filling the pores of the formed carbide skeleton, is provided so that the surplus of silicon or silicon alloy will form a surface coating of at least one surface of the manufactured diamond composite.
- 25 10. Use of the method according to claim 1 for forming a substrate for a diamond film.
11. A diamond composite manufactured by the method according to claim 1, wherein one side of the composite is coated with a thin layer of aluminium nitride.

ABSTRACT OF THE DISCLOSURE

The present invention relates to a method for manufacturing a diamond composite, comprising the steps of: a) mixing diamonds with additives, the mixture comprising at least 50 wt% and less than 95 wt% of diamonds and more than 5 wt% additives; b) forming a work piece from the mixture using a pressure of at least 100 Mpa; c) heating the formed work piece to at least 300°C for removing possible water and wholly or partially removing additives; d) heating the work piece and controlling the heating temperature and heating time so that a certain desired amount of graphite is created by graphitization of diamonds, wherein the amount of graphite created by graphitization is 3-50 wt% of the amount of diamond; e) infiltrating silicon or silicon alloy into the work piece, the infiltration of silicon or silicon alloy being performed at a temperature below 1900°C and a pressure less than 50 bars; and f) heating the infiltrated work piece to form silicon carbide, or other carbides, (and silicides), thereby creating a final diamond composite.

INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP2004/050367

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C04B35/573 C04B41/50 C04B35/52 C22C26/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C04B C22C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EP0-Internal

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 6 179 886 B1 (DANCHUKOVA LIJA VLADIMIROVNA ET AL) 30 January 2001 (2001-01-30) cited in the application example 1	1-10
A	WO 00/18702 A (FRENTON LTD ; SCIENT RESEARCH CENTER AMT OF (RU) ; GORDEEV SERGEY K (RU) 6 April 2000 (2000-04-06) cited in the application example 1	1-10
A	US 6 447 852 B1 (DANCHUKOVA LIJA VLADIMIROVNA ET AL) 10 September 2002 (2002-09-10) example 1	1-10

☒ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search

9 August 2004

Date of mailing of the international search report

2004 08 09

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Sala, P

INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP2004/050367

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 4 453 951 A (OHNO JOHN M) 12 June 1984 (1984-06-12) column 3, line 24 - column 4, line 17; claim 1 -----	1-10

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP2004/050367

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 6179886	B1	30-01-2001	
		RU 2147508 C1	20-04-2000
		RU 2131805 C1	20-06-1999
		RU 2151126 C1	20-06-2000
		RU 2147509 C1	20-04-2000
		AU 8975798 A	29-03-1999
		BR 9811633 A	26-09-2000
		CN 1125793 B	29-10-2003
		WO 9912866 A1	18-03-1999
		EP 1019337 A1	19-07-2000
		AT 224858 T	15-10-2002
		AU 749202 B2	20-06-2002
		CA 2301775 A1	18-03-1999
		CZ 20000613 A3	12-12-2001
		DE 69808324 D1	31-10-2002
		DE 69808324 T2	22-05-2003
		EA 3437 B1	24-04-2003
		EP 1253123 A1	30-10-2002
		ES 2187055 T3	16-05-2003
		HK 1030767 A1	02-07-2004
		JP 2001515836 T	25-09-2001
		PL 339012 A1	04-12-2000
		AT 229486 T	15-12-2002
		AU 9623098 A	29-03-1999
		BR 9811635 A	08-08-2000
		CA 2301611 A1	18-03-1999
		CN 1272100 T	01-11-2000
		CZ 20000724 A3	12-12-2001
		DE 69810141 D1	23-01-2003
		DE 69810141 T2	11-03-2004
		EA 1843 B1	27-08-2001
		WO 9912867 A1	18-03-1999
		EP 1019338 A1	19-07-2000
		ES 2190814 T3	16-08-2003
		JP 2001515927 T	25-09-2001
		PL 339060 A1	04-12-2000
WO 0018702	A	06-04-2000	
		RU 2147982 C1	27-04-2000
		RU 2151814 C1	27-06-2000
		AT 218520 T	15-06-2002
		AU 759804 B2	01-05-2003
		AU 3811699 A	17-04-2000
		BR 9914094 A	12-06-2001
		CA 2342986 A1	06-04-2000
		CN 1320108 T	31-10-2001
		DE 69901723 D1	11-07-2002
		DE 69901723 T2	13-02-2003
		EA 3715 B1	28-08-2003
		WO 0018702 A1	06-04-2000
		EP 1117625 A1	25-07-2001
		ES 2182517 T3	01-03-2003
		JP 2002525262 T	13-08-2002
		US 6709747 B1	23-03-2004
US 6447852	B1	10-09-2002	
		US 6709747 B1	23-03-2004
US 4453951	A	12-06-1984	
		AU 7919782 A	29-07-1982
		EP 0056945 A1	04-08-1982
		JP 57135780 A	21-08-1982

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP2004/050367

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 4453951	A	ZA 8200101 A	24-11-1982
		AT 13046 T	15-05-1985
		AU 7255381 A	14-01-1982
		BR 8108689 A	25-05-1982
		CA 1171249 A1	24-07-1984
		DE 3170267 D1	05-06-1985
		EP 0043542 A1	13-01-1982
		JP 57501027 T	10-06-1982
		WO 8200138 A1	21-01-1982
		US 4428755 A	31-01-1984
		ZA 8104403 A	27-10-1982
